# NEW IMPROVED EQUATIONS FOR Na/K AND SiO<sub>2</sub> GEOTHERMOMETERS BY ERROR PROPAGATION

Surendra Pal VERMA 1,2 and Edgar SANTOYO 1

<sup>1</sup> Depto. de Geotermia, IIE, Apartado Postal 1-475, Cuernavaca, Mor. 62001, Mexico. <sup>2</sup> (present address) Lab. Energia Solar, IIM-UNAM, A. P. 34, Temixco, Mor. 62580, Mexico.

**Key words:** Geothermometer, error propagation, geothermometric equation, geothermal exploration

#### Abstract

We present here new improved equations for the Na/K and SiO<sub>2</sub> geothermometers, obtained by the theory of error propagation and statistical treatment of the data base used by Fournier (1979) and Fournier and Potter II (1982). The minimum and "reasonable" maximum errors associated with the use of the new geothermometric equation for Na/K range respectively from ~29°C—54°C and ~31°C—57°C for the temperature range of 150°C—350°C. These values are significantly lower than ~45°C-82°C and ~46°C-84°C found for the original Fournier equation. Similarly, the new equations for the SiO<sub>2</sub> geothermometer, for 20°C to 310°C, result in the minimum and "reasonable" maximum errors of ~0.5°C—4°C -1.8°C-18°C respectively, which are much lower than -1.7°C-19°C and -2.3°C-55°C for the original regression. Finally, an application of the Na/K geothermometer to actual field data shows a better performance of our new equation.

## 1. INTRODUCTION

Without exception, a physical or a chemical parameter measured by any experimental technique has an associated experimental or analytical error (probable error), whether systematic (resulting from faulty instrument, its calibration or from bias on behalf of the experimenter) or random (that make the result different from the "true" value with a reproducible discrepancy).

Data reduction and error analysis have been fundamental in physical sciences (e.g., Bevington, 1969). The evaluation and treatment of these errors have also become a routine matter in some branches of Earth Sciences, such as isotope geology and geochronology (e.g., York, 1967, 1969; Lanphere and Dalrymple, 1967; Brooks *et al.*, 1972; Russell, 1977; Faure, 1986), major element geochemistry and modelling (e.g., Bryan *et al.*, 1968; Wright and Doherty, 1970; Verma *et al.*, 1991), trace element geochemistry (e.g., Kleeman, 1967; Ingamells, 1974; Verma and Schilling, 1982), and compilation of international geochemical reference samples (e.g., Govindaraju, 1980; Gladney and Burns, 1983; Abbey, 1986; Velasco and Verma, 1993).

In geothermal research involving the application of geothermometers to hydrothermal systems, such criteria have not been generally applied (e.g., Buntebarth, 1984; Henley *et al.*, 1984). It is only very recently that the geothermometers have been looked at from the standpoint of deviation between a regression curve and the actual data (Nieva and Nieva, 1987) or totai propagated analytical and regression errors (Santoyo and Verma, 1991, 1993).

In the present work, we derive newly revised and improved geothermometric equations for two chemical geothermometers (Na/K and SiO<sub>2</sub>), with significantly reduced total propagated errors. This has been achieved through statistical tests of the data population and application of the theory of error propagation to each of these geothermometers. We report here these new equations and compare their performance with the original ones proposed by Fournier (1979) and Fournier and Potter II (1982). Our choice of these two rather old geothermometric equations is based on the fact that these are still among the most widely used equations in geothermal exploration.

## 2. PREVIOUS WORK

There are several sources of errors or uncertainties associated with the use of chemical geothermometers, such as, errors of the regression coefficients of the geothermometric equations, accuracy and precision of the analytical determinations of the chemical species in a given sample, sampling errors, calibration errors of the geothermometers for high temperature hydrothermal systems, and errors related to the geologic and thermodynamic processes of the chemical equilibria involved in the reactions (Santoyo and Verma, 1993). These different errors affect the interpretation of the final predicted temperatures and are propagated according to certain rules summarized by Bevington (1969) in the chapter on propagation of errors of his book. Because several aspects related to the geologic and thermodynamic processes have been discussed by Henley *et al.* (1984) and Giggenbach (1988), we have not included them in our present paper.

As a first step, we have evaluated the effect of analytical errors in Na/K and  $SiO_2$  geothermometers (Santoyo and Verma, 1991). We have expanded this work by combining the analytical errors with those associated with the regression analysis of the geothermometric equations (Santoyo and Verma, 1993).

## 3. METHODOLOGY

We describe very briefly the procedure we have followed for evaluating the two geothermometers (Na/K and SiO<sub>2</sub>) and proposing new improved equations.

## 3.1 Na/K geothermometer

The Na/K geothermometer proposed by Fournier (1979) on the basis of well data from many locations around the world (n = 36) is:

$$t = \frac{A_1}{\log(Na/K) + A_2} - 273.15 \tag{1}$$

where t = temperature in  ${}^{\circ}$ C, Na and K are the concentration of these cations in ppm and n is the number of the available data employed in the regression. The regression coefficients of eq. (1) are: A, = 1217  $\pm \sigma_{A1}$  and A, = 1.483  $\pm \sigma_{A2}$ , where  $\sigma_{A1}$  = 93.9 and  $\sigma_{A2}$  = 0.2076 are the standard deviations of these regression coefficients (Santoyo and Verma, 1993).

The propagated combined error  $(\sigma_t)$  in the temperature t for this equation is (Santoyo and Verma, 1993):

$$\sigma_t = \frac{A_1}{\log(Na/K) + A_2}.$$

$$\left[\frac{\sigma_{A1}^{2}}{A_{1}^{2}} + \frac{1}{(\ln 10)^{2}} \cdot \left[\frac{\sigma_{Na}^{2}}{Na^{2}} + \frac{\sigma_{K}^{2}}{K^{2}}\right] + \sigma_{A2}^{2}\right]^{1/2}$$

$$(2)$$

A limiting case for estimating the minimum error of this geothermometer ( $\sigma_{t,min}$ ) will be when the analytical errors of sample determination  $\sigma_{Na} = \sigma_K = 0$ . The above equation reduces to:

$$\sigma_{t-min} = \frac{A_1}{\log(Na/K) + A_2} \cdot \frac{\left[\sigma_{A1}^2 + \frac{\sigma_{A2}^2}{A_1^2} + \frac{\sigma_{A2}^2}{(\log(Na/K) + A_2)^2}\right]^{1/2}}{\left[\sigma_{A1}^2 + \frac{\sigma_{A2}^2}{A_1^2} + \frac{\sigma_{A2}^2}{(\log(Na/K) + A_2)^2}\right]^{1/2}}$$
(3)

Santoyo and Verma (1993) used eq. (2) and (3) to evaluate this geothermometer (eq. 1). Here we use them to evaluate our new equation (eq. 7 in the "RESULTS" section), derived by minimizing the errors associated with the regression coefficients of the existing equations.

Our procedure is to first estimate the average deviation  $(1\sigma)$  of all the data points with respect to the originally proposed geothermometric equation (eq. 1). Then we use a statistical criterion of rejecting all those data points lying outside:

the  $\pm$  4 $\sigma$  deviation from the original equation, derive the new geothermometric equation using a BMDPAR subroutine (Dixon, 1981) and the new average 4 $\sigma$  deviation, and repeat this procedure until no more data points are rejected (4 $\sigma$  method).

Thus we have obtained a new equation for this geothermometer (eq. 7 in the "RESULTS" section). The total propagated errors of this new equation are then compared with those associated with the originally proposed geothermometric equation (eq. 1).

## 3.2 SiO<sub>2</sub> geothermometer

The  $SiO_2$  geothermometer is based on the regression of experimental data of silica solubility and temperature (n = 32; Fournier and Potter II, 1982):

$$t = C, + C_{5}S + C_{3}S^{2} + C_{4}S^{3} + C_{5}\log S$$
 (4)

where  $t = \text{temperature in }^{\circ}\text{C}$  in the range of 20°C to 330°C, S is the concentration of  $\text{SiO}_2$  in a sample in mg/kg (ppm), and Ci are the regression coefficients as follows (Santoyo and Verma, 1993):

$$\begin{array}{lll} C_1 = -42.1981 & \pm \sigma_{C1} & (= 1.3454) \\ C_2 = & 0.288313 & \pm \sigma_{C2} & (= 1.337 \times 10^{-2}) \\ C_3 = -3.6686 \times 10^{-4} & \pm \sigma_{C3} & (= 3.152 \times 10^{-5}) \\ C_4 = & 3.1665 \times 10^{-7} & \pm \sigma_{C4} & (= 2.421 \times 10^{-8}) \\ C_5 = & 77.034 & \pm \sigma_{C5} & (= 1.21637) \end{array}$$

At first sight, Fournier and Potter II (1982) fit of the regression equation to the experimental data appears to be perfect. However, the associated errors in the regression coefficients, particularly in the higher order terms (quadratic and cubic), are very large (8.6% and 7.6% respectively), as is seen from the above error data estimated by Santoyo and Verma (1993). Furthermore, such a complicated regression (eq. 4) and a "perfect" fit to the data points implicitly and perhaps erroneously assumes that the data of the silica solubility experiment have *no* analytical errors.

The propagated error  $(\sigma_i)$  of eq. (4) can be estimated by the general equation:

$$\sigma_{t} = [\sigma_{C1}^{2} + (S^{2}\sigma_{C2}^{2} + C_{2}^{2}\sigma_{S}^{2}) + (S^{4}\sigma_{C3}^{2} + 4C_{3}^{2}S^{2}\sigma_{S}^{2}) + (S^{6}\sigma_{C4}^{2} + 9C_{4}^{2}S^{4}\sigma_{S}^{2}) + ((\log S)^{2}\sigma_{C5}^{2} + (C_{5}^{2}/(\ln 10)^{2})(\sigma_{S}^{2}/S^{2}))]^{1/2}$$
(5)

and the minimum error  $(\sigma_{t-min})$  is given when  $\sigma_s = 0$ :

$$\sigma_{\text{t-min}} = [\sigma_{\text{Cl}}^2 + S^2 \sigma_{\text{C2}}^2 + S^4 \sigma_{\text{C3}}^2 + S^6 \sigma_{\text{C4}}^2 + (\log S)^2 \sigma_{\text{C5}}^2]^{1/2}$$
 (6)

In order to obtain new improved equations for this geothermometer, we have attempted to fit:

- (i) a simpler equation to the silica solubility data (Fournier and Potter II, 1982) in the temperature range of 20°C—210°C, and
- (ii) a linear equation for the data from 210°C to 330°C.

Although the  $SiO_2$  geothermometer is based on controlled laboratory experimental data, rather than the worldwide well data of the Na/K geothermometer, we have applied the same strict statistical criterion of rejecting only those data points lying outside the  $\pm 4\sigma$  deviation from the geothermometric equation, derive the new equation and the new average  $4\sigma$  deviation, and repeat this procedure until no more data points are rejected ( $4\sigma$  method).

Thus, we have obtained two new equations (eq. 8 and 9 in the next section) for this geothermometer applicable for temperatures between 20°C—210°C and >210°C—310°C. The total propagated errors of these new equations are then derived according to the theory of error propagation and the results are compared with the originally proposed geothermometric equation (eq. 4).

## 4. RESULTS

## 4.1 Na/K geotherniometer

The data points used by Fournier (1979) as well as his geothermometric equation are plotted in Figure 1. The new

equation after rejecting the "outlier" data is also shown in Figure 1 for the  $4\sigma$  method.

The new equation is  $(n = 33 \text{ for the } 4\sigma \text{ method})$ :

$$t = \frac{1289(\pm 65)}{\log(Na/K) + 1.635(\pm 0.145)} -273.15 \tag{7}$$

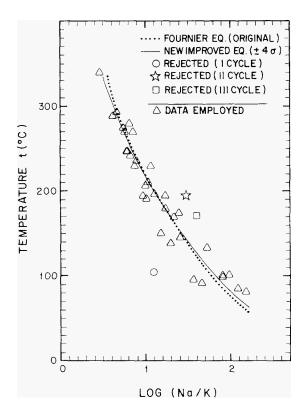


Figure 1. Log(Na/K) — temperature plot for the data used in the Na/K geothermometer. The original equation (Fournier, 1979) proposed on the basis of these data is shown by dotted curve. The continuous curve is the new geothermometric equation for the  $4\sigma$  method. Three data were successively eliminated as "outliers", these rejected data are also shown for reference.

We have evaluated the errors associated with the use of this new equation (eq. 7). Figure 2 compares them with the original equation (Fournier, 1979; Santoyo and Verma, 1993). For temperatures ranging from 150°C to 350°C, our results show that the minimum total propagated error ( $\sigma_{t-min}$ ) of the new geothermometric equation vanes from -29°C to  $\sim 54$ °C. These values are significantly lower than for the original Fournier equation for which we found minimum error values from -45°C to 82°C (Figure 2). The total propagated errors ( $\sigma_t$ ) for the new equation are also much lower ( $\sim 31$ °C—57°C) than the Fournier (1979) regression ( $\sim 46$ °C—84°C; Figure 2).

Thus, the minimum total propagated error ( $\sigma_{\text{t-min}}$ ) eq. 3) of the new equation for the temperature range of 150°C to 350°C varies from 15% to 20%, which is considerably lower than the corresponding error (23% to 30%) of the original Fournier equation (eq. 1). Similarly, the "reasonable" maximum error ( $\sigma_{\text{t}}$ ; eq. 2) associated with the use of the new geothermometer for the same temperature range is 16%—20%, much lower than 24%—30% estimated for the original Fournier equation.

The propagated analytical errors for the hypothetical case of no errors in the regression coefficients ( $\sigma_{A1}=\sigma_{A2}=0$ ) are also shown in Figure 2. These errors are relatively small (<10°C) for the case of  $\sigma_{Na}=\sigma_{K}=5\%$  but can reach somewhat higher values (-20°C) for  $\sigma_{Na}=\sigma_{K}=10\%$ . However, note that our new equation still results in somewhat lower error values (vs and vs' curves always lie below the f and f' curves; Figure 2) throughout the temperature range.

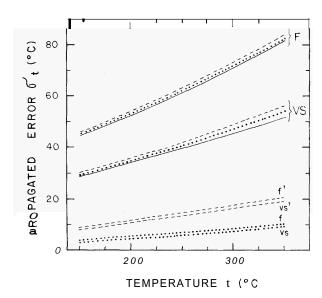


Figure 2. Predicted temperature (t) — propagated error  $(\sigma_0)$  plot for the Na/K geothermometer. F, f, and f' = Fournier (1979); VS, vs, and vs' = this work. Upper curves (grouped as F and VS) are for total propagated errors of the regression coefficients ( $\sigma_{A1}$  and  $\sigma_{A2}$ ) and the analytical errors ( $\sigma_{Na}$  and  $\sigma_{K}$ ) (dotted curves here are for  $\sigma_{Na}$  =  $\sigma_{K}$  = 5% and dashed curves for  $\sigma_{Na}$  =  $\sigma_{K}$  = 10%, whereas continuous curves represents  $\sigma_{t-min}$ ). Lower curves (denominated f, vs, f', and vs') are for propagated analytical errors, assuming no errors in the regression coefficients ( $\sigma_{A1}$  =  $\sigma_{A2}$  = 0). The curves f and vs are for  $\sigma_{Na}$  =  $\sigma_{K}$  = 5% (dotted curves) and f' and vs' for  $\sigma_{Na}$  =  $\sigma_{K}$  = 10%(dashed curves).

# 4.2 SiO<sub>2</sub> geothermometer

The data points (n = 32) used by Fournier and Potter II (1982) and their geothermonietric equation are plotted in Figure 3.

The first new geothermometric equation (n = 20) is:

$$t = C_1 + C_2 S + C_3 S^2 + C_4 \log S$$
 (8)

where  $t = \text{temperature in }^{\circ}\text{C}$  in the range of 20°C to 210°C, S is the concentration of SiO<sub>2</sub> in a sample in mg/kg (ppm), and Ci are the regression coefficients as follows:

$$C_1 = -44.119$$
  $\pm \sigma_{C1}$  (= 0.438)  
 $C_2 = 0.24469$   $\pm \sigma_{C2}$  (= 0.00573)  
 $C_3 = -1.7414 \times 10^4$   $\pm \sigma_{C3}$  (= 1.365×10<sup>-5</sup>)  
 $C_4 = 79.305$   $\pm \sigma_{C4}$  (= 0.427)

The 40 method did not reject any data.

# VERMA and SANTOYO

The second is a linear equation for the remaining data from  $210^{\circ}$ C to  $330^{\circ}$ C (two highest temperature data were successively eliminated by the  $4\sigma$  method; the applicable temperature range is therefore  $>210^{\circ}$ C— $310^{\circ}$ C). This second equation for the SiO<sub>2</sub> geothermometer (n = 11) is as follows:

$$t = C_1 + C_2 S \tag{9}$$

The regression coefficients are as follows:

$$C_{2} = 140.82$$
  $\pm \sigma_{C1} (= 0.00)$   $\pm \sigma_{C2} (= 0.00179)$ 

The total propagated error  $(\sigma_1)$  of eq. (8) can be estimated by:

$$\sigma_{t} = [\sigma_{C1}^{2} + (S^{2}\sigma_{C2}^{2} + C_{2}^{2}\sigma_{S}^{2}) + (S^{4}\sigma_{C3}^{2} + 4C_{3}^{2}S^{2}\sigma_{S}^{2}) + ((\log S)^{2}\sigma_{C4}^{2} + (C_{4}^{2}/(\ln 10)^{2})(\sigma_{S}^{2}/S^{2}))]^{1/2}$$
(10)

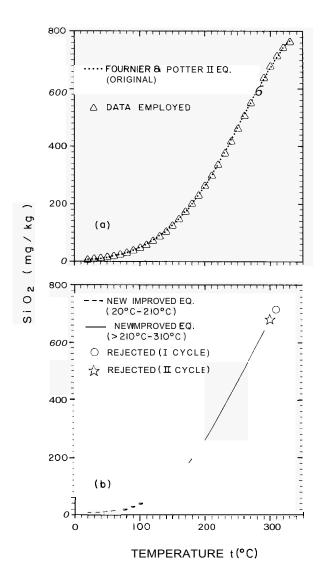


Figure 3.  $SiO_2$  concentration (mg/kg) — temperature (t) plot for the data points used in the  $SiO_2$  geothermometer. (a) The original equation (Fournier and Potter II, 1982) is shown by dotted curve. (b) The dashed and continuous curves are the new geothermometric equations (eq. 8 and 9 obtained by the  $4\sigma$  method) for the temperature range of  $20^{\circ}$ C— $210^{\circ}$ C and  $>210^{\circ}$ C— $310^{\circ}$ C respectively.

and the minimum error  $(\sigma_{t-min})$  is given when  $\sigma_s = 0$ :

$$\sigma_{\text{t-min}} = [\sigma_{\text{Cl}}^2 + S^2 \sigma_{\text{C2}}^2 + S^4 \sigma_{\text{C3}}^2 + (\log S)^2 \sigma_{\text{C4}}^2]^{1/2}$$
 (11)

Similarly, the total propagated error  $(\sigma_i)$  of eq. (9) can be estimated by:

$$\sigma_{t} = [S^{2}\sigma_{C2}^{2} + C_{2}^{2}\sigma_{S}^{2}]^{1/2}$$
(12)

and the minimum error  $(\sigma_{t-min})$  is given when  $\sigma_s = 0$ :

$$\sigma_{\text{t-min}} = S\sigma_{C2} \tag{13}$$

We have evaluated the errors associated with the use of these new equations (8 and 9) and compared them with the performance of the original eq. (4) by Fournier and Potter II (1982) in Figure 4 (a and b).

For temperatures ranging from 20°C to 310°C, our results show that the minimum error  $(\sigma_{\text{t-min}})$  of the new geothermometric equations varies from ~0.5°C to -4°C. These values are considerably smaller than for the original Fournier and Potter II equation, for which we found minimum error values from ~1.7°C to ~19°C (Figure 4a). The total propagated-errors  $(\sigma_t)$  for the new equations (for analytical error in SiO<sub>2</sub>,  $\sigma_s$  from 5% to 10%) are also much lower (-1.8°C—18°C) than, the original regression  $(\sim 2.3$ °C—55°C; Figure 4a).

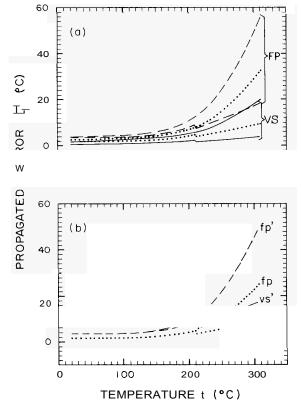


Figure 4. Predicted temperature (t) —propagated error  $(\sigma_l)$  plot for the SiO<sub>2</sub> geothermometer. Dotted curves refer to the case of  $a_r = 5\%$  and dashed curves are for  $\sigma_s = 10\%$ . (a) Total propagated error in the regression coefficients  $(\sigma_{Ci})$  and the analytical error  $(\sigma_s)$ . Continuous curves are for  $\sigma_{t\text{-min}}$ . VS = Verma and Santoyo (this work); FP = Fournier and Potter II. (b) Propagated analytical error, assuming no error in the regression coefficients  $(\sigma_{Ci} = 0)$  [fp  $(\sigma_s = 5\%)$  and fp'  $(\sigma_s = 10\%)$  = Fournier and Potter II; vs and vs' = this work).

Thus, the minimum errors ( $\sigma_{\text{t-min}}$ ) of the new equations (eq. 8 and 9) for the temperature range of 20°C to 310°C vary from 0.8% to 2.8%, which are considerably lower than the corresponding error (2.2% to 8.3%) of the original Fournier and Potter II equation (eq. 4). Similarly, the "reasonable" maximum errors ( $\sigma$ ) associated with the use of the new geothermometers for the same temperature range are 1.7% to 17.8%, which are consistently lower than 2.7% to 18.8% estimated for the original equation.

The propagated analytical errors for the hypothetical case of no errors in the regression coefficients ( $\sigma_{\rm Ci}=0$ ) are shown in Figure 4b. These errors are relatively small ( $<-9^{\circ}{\rm C}$  for the new equations and  $<25^{\circ}{\rm C}$  for the original one) for the case of a, = 5%, but can reach somewhat higher values (up to  $-18^{\circ}{\rm C}$  for the new equations and  $-55^{\circ}{\rm C}$  for the original one) for  $\sigma_{\rm S}=10\%$ .

## 5. APPLICATION TO ACTUAL FIELD DATA

In order to further evaluate the performance of our new improved equation (eq. 7) for the Na/K geothermometer in comparison with the original equation (eq. 1), we have computed the corresponding temperatures from the chemical data (compiled by Nieva and Nieva, 1987) from hot springs and wells in Mexico (Cerro Prieto) and other countries. In Figure 5, the relative % temperature deviation  $\{100 \text{ *} (t_{vs} - t_f)/t_m\}$  of the new and old equations is plotted as a function of the measured temperature (t<sub>m</sub>). The negative % deviation values are for samples for which the temperature (f,) predicted by the new equation (eq. 7) is closer to the measured temperature (t<sub>m</sub>). On the other hand, the positive % deviation values correspond to samples for which the temperature (t,) by original equation (eq. 1) falls closer to the measured temperature (t<sub>m</sub>). We conclude that the temperatures predicted by the new equation are in general closer to the measured temperatures, because most of the data fall in the negative % deviation field in Figure 5.

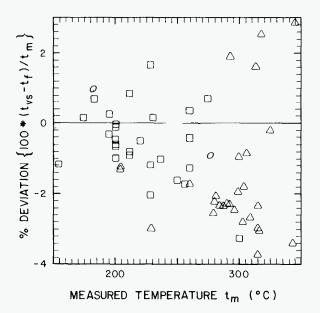


Figure 5. A plot of % deviation  $\{100 \cdot (t_{vs} - t_f)/t_m\}$  as a function of the measured temperature  $(t_m)$  in hot springs and wells  $(t_{vs} = temperature computed from the new equation and t, = temperature from the original Fournier equation). The symbols are: triangles for Mexico (Cerro Prieto) and squares for other countries.$ 

We have not attempted to evaluate from field data the performance of our new equations (eq. 8 and 9) for the  $SiO_2$  geothermometer. This is because, in the lower temperature range of  $20^{\circ}C$ — $210^{\circ}C$ , the new (eq. 8) and original (eq. 4) equations should give very similar temperature values, as there were no "outliers" in this part of the regression (see Figure 4). For the higher temperature range (> $210^{\circ}C$ — $310^{\circ}C$ ), the new equation (eq. 9), because of its simplicity, is likely to be more useful in comparison with the original equation (linear regression of eq. 9 *versus* the original more complicated eq. 4).

## 6. DISCUSSION

We have clearly demonstrated that the existing geothermometric equations for the Na/K and  $SiO_2$  geothermometers result in very large total propagated errors which are perhaps unacceptable in modern geothermal exploration. We have also presented our new improved equations for these geothermometers and shown by the application of the theory of error propagation that they should perform much better than Fournier (1979) and Fournier and Potter II (1982) equations. In fact, we have applied both sets of equations—the new as well as the original equation for the Na/K geothermometer—to actual field data, which once again confirms that our new equation performs better than the original.

In future work, we plan to apply this methodology to chemical geothermometers of Fouillac and Michard (1981) and Fournier (1990), and to gas geothermometers (CO<sub>2</sub>, H<sub>2</sub>S and H<sub>2</sub>) proposed by Arnórsson and Gunnlaugsson (1985), in order to evaluate the associated total propagated errors. We will also attempt to obtain corresponding revised and improved equations with considerably reduced, errors and evaluate their performance with actual field data.

This approach **is** likely to provide an additional tool to the field geologist or geochemist, by which he/she will be able to make better decisions regarding the temperature field distribution in an area under exploration. Furthermore, this kind of work shows an entirely new way of explaining, at least partly, the discrepancies in the temperatures predicted by different geothermometers.

Such an approach will be particularly useful in geothermal exploration when combined with other recent techniques of temperature field simulations from cooling of a magma chamber for the entire volcanic history of the geothermal areas (Verma and Andaverde, in press, this volume).

# 7. CONCLUSIONS

We have clearly demonstrated that our new equations (eq. 7 for the Na/K and eq. 8 and 9 for the  $SiO_2$  geothermometers) result in very significantly reduced propagated errors, in comparison with the originally proposed equations for these two geothermometers by Fournier (1979) and Fournier and Potter II (1982). Further, from actual field data we have also shown that in most cases the new equation predicts temperatures closer to the actually measured values than the original Na/K geothermometer. We therefore strongly recommend the use of these improved geothermometric equations for geothermal exploration.

We have also shown that the application of the theory of error propagation in geothermal research is a novel and rewarding scientific and technical approach. This could eventually lead to a significant progress in the development of new techniques for

## VERMA and SANTOYO

geothermal exploration, particularly when employed in conjunction with other recent developments of temperature field simulations from magma chambers (Verma and Andaverde, in press, this volume).

### 8. ACKNOWLEDGEMENTS

We are grateful to Robert H. Mariner and Mike Thompson for comments on an earlier version of this paper.

## 9. REFERENCES

Abbey, S. (1986). A study of robust estimates. *Geostand. Newslett.*, Vol. 10, pp. 159-168.

Arnórsson, S. and Gunnlaugsson, H. (1985). New gas geothermometers for geothermal exploration — Calibration and application. *Geochim. Cosmochim. Acta*, Vol. 49. pp. 1307-1325.

Bevington, P.R. (1969). Data reduction and error analysis for the physical sciences. Mc-Graw Hill Book Co., New York, 336 pp.

Brooks, C., Hart, S.R. and Wendt, I. (1972). Realistic use of two-error regression treatments as applied to rubidium-strontium data. *Rev. Geophys. Space Phys.*, Vol. 10, pp. 551-577.

Bryan, W.B., Finger, L.W. and Chayes, F. (1968). Estimating proportions in petrographic mixing equations by least-squares approximation. *Science*. Vol. 163, pp. 926-927.

Buntebarth, G. (1984). Geothennics. An introduction. Springer-Verlag, Berlin, 144 pp.

Dixon, W.J. (1981). *BMDP, Statistical Software*. Dept. of Biomathematics, Los Angeles, Univ. of California Press, 335 pp.

Faure, G. (1986). *Principles of* isotope *geology*. Second Edition, Wiley, New York, 653 pp.

Fouillac, C. and Michard, G. (1981). Sodium/lithium ratio in water applied to geothermometry of geothermal reservoirs. *Geothemics*, Vol. 10, pp. 55-70.

Fournier, R.O. (1979). A revised equation for the Na/K geothermometer. *Geothem. Res. Coun. Trans.*, Vol. 3, pp. 221-224.

Fournier, R.O. (1990). The interpretation of Na-K-Ca relations in geothermal waters. *Geothem. Res. Coun. Trans.*, Vol. 14, pp. 1421-1425.

Fournier, R.O. and Potter II, R.W. (1982). A revised and expanded silica (quartz) geothermometer. *Geothem. Res. Coun. Bull.*, Vol. 11, pp. 3-12.

Giggenbach, W.F. (1988). Geothermal solute equilibria. Derivation of Na-K-Mg-Ca geoindicators. *Geochim. Cosmochim. Acta*, Vol. 52, pp. 2749-2765.

Gladney, E.S. and Burns, C.E. (1983). 1982 compilation of elemental concentrations in eleven United States Geological Survey rock standards. *Geosrand. Newslett.*, Vol. 7, pp. 3-226.

Govindaraju, K. (1980). Report (1980) on three GIT-IWG rock reference samples: anorthosite from Greenland, AN-G; basalte d'Essey-la-Cote, BE-N; granite de Beauvoir, MA-N. *Geostand. Newslett.*, Vol. 4, pp. 49-138.

Henley, R.W., Truesdell, A.H., Barton Jr., P.B. and Whitney, J.A. (1984). *Fluid-mineral equilibria in hydrothermal systems*. Economic Geology Publishing Company, El Paso, Texas, 267 pp.

Ingamells, C.O. (1974). New approaches to geochemical analysis and sampling. *Talanta*, Vol. 21, pp. 141-155.

Kleeman, A.W. (1967). Sampling error in the chemical analysis of rocks. *J. Geol. Soc. Australia*, Vol. 14, pp. 43-48.

Lanphere, M.A. and Dalrymple, G.B. (1967). K-Ar and Rb-Sr measurements on P-207, the USGS interlaboratory standard muscovite. *Geochim. Cosmochim. Acta*, Vol. 31, pp. 1091-1094.

Nieva, D. and Nieva, R. (1987). Developments in geothermal energy in Mexico-Part twelve. **A** cationic geothermometer for prospecting of geothermal resources. *Heat Rec. Sys.*, Vol. 7, pp. 243-258.

Russell, R.D. (1977). A solution in closed form for the isotopic dilution analysis of strontium. *Chem. Geol.*, Vol. 20, pp. 307-314.

Santoyo, E. and Verma, S.P. (1991). Evaluación de errores en **el** uso de geotermómetros quimicos para la prospección de recursos geotérmicos. *Actas Fac. Ciencias Tierra UANL Linares*, Vol. 6, pp. 5-10.

Santoyo, E. and Verma, S.P. (1993). Evaluación de errores en el uso de los geotermometros de SiO<sub>2</sub> y Na/K para la determinación de temperaturas en sistemas geotérmicos. *Geofs. Int.*, Vol. 32, pp. 287-298.

Velasco, F. and Verma, S.P. (1993). Aplicación del sistema GEOBAS al análisis estadistico de elementos de las tierras raras en muestras internacionales de referencia geoquímica. *Geofis. Int.*, Vol. 32, pp. 209-219.

Verma, S.P. and Andaverde, J. (in press). Temperature field distribution from cooling of a magma chamber. *Proc. World Geothem. Congr. 1995* (this volume).

Verma, S.P. and Schilling, J.-G. (1982). Galapagos hot spot-spreading center system. 2. <sup>87</sup>Sr/<sup>86</sup>Sr and large lithophile element variations (85°W-101°W). *J. Geophys. Res.*, Vol. 87, pp. 10838-10856.

Verma, S.P., Carrasco-Núñez, G. and Milán, M. (1991). Geology and geochemistry of Amealco caldera, Qro., Mexico. *J. Volcanol. Geotherm. Res.*, Vol. 47, pp. 105-127.

Wright, T.L. and Doherty, P.C. (1970). A linear programming and least squares computer method for solving petrologic mixing problems. *Geol. Soc. Am. Bull.*, Vol. 81, pp. 1995-2008.

York, D. (1967). The best isochron. *Earth Planet. Sci. Lett.*, Vol. 2, pp. 479-482.

York, D. (1969). Least-squares fitting of a straight line with correlated errors. *Earth Planet. Sci. Lett.*, Vol. 5, pp. 320-324.