

A HYDROGEOCHEMICAL MODEL OF THE TONGONAN GEOTHERMAL FIELD

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ABSTRACT

Chemistry and enthalpy data have been used to identify the Upper Mahiao reservoir fluids as the most mineralized high temperature region of the Tongonan field so far drilled. Broad cooling and dilution trends have been delineated together with directions of preferential steam migration. The response of discharge enthalpy and fluid chemistry to changing wellhead pressure is described and its importance in chemical interpretation is stressed. The permeability of structural features is discussed and the possibility of some structural control on hydrology is considered.

PROJECT HISTORY

The Tongonan Geothermal Field is located on the island of Leyte in the Republic of the Philippines and is associated with the Philippine Fault Zone which strikes northwest through the island arc system.

Exploration drilling began in 1974 with shallow temperature gradient wells sited on the basis of surface geology and distribution of major surface manifestations, particularly hot chloride water discharging from the Bao Springs. Deep drilling commenced in 1977 with well 401, the first well to tap the deep high temperature (over 300°C) chloride water that is typical of the Tongonan geothermal reservoir. Figure 1 shows the distribution of wells in the Tongonan field as of early 1982. The Lower Mahiao and Sambaloran sectors will supply steam to the 112.5 MW Leyte Power Station in late 1982 and currently the Malitbog sector is being developed for a second station.

GEOLOGY OF THE TONGONAN FIELD

At Tongonan andesitic volcanics plus minor marine sediments are found to 2 km below sea level. The andesite is intruded in the central part of the field by a diorite body. The topography of the upper surface of this diorite pluton, which has been closely defined by drilling is shown in Figure 1 (Palma 1981). The contact between the andesite and diorite pluton comprises andesites, microdiorite/diorite dykes and andesite/diorite breccias. This has been called the Transition Zone and it has permeability which is considered to be associated with the emplacement of the pluton.

The Philippine Fault Zone with conjugate minor faults has contributed significantly to the present-day structure of the field.

GENERAL GEOCHEMISTRY

Table 1 lists chemical data for Tongonan well waters separated at atmospheric pressure and for steam separated at higher pressures. The data are those collected under throttled, high wellhead pressure conditions when the total discharge enthalpy is generally lowest. The discharge waters are typically neutral pH sodium chloride brines with accompanying potassium and Significant calcium and boron. Non-condensable gas concentrations vary considerably throughout the field with CO₂ ranging between 40 and 800 mmol/100 moles in the total discharge.

The Bao Valley neutral chloride springs are the most impressive surface manifestations in the Tongonan field and wells TGE 4 and 5 intersected the same fluid at shallow depth. The chemical

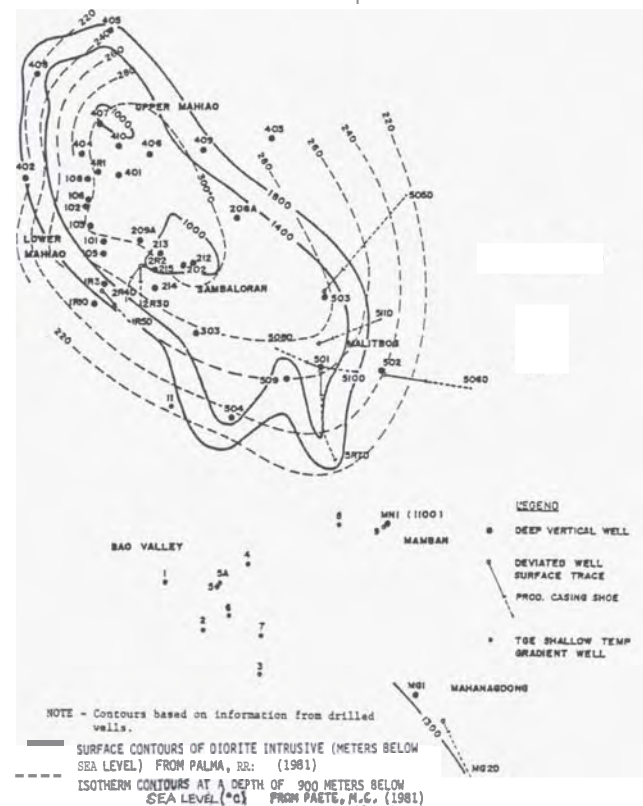


Figure 1: Tongonan Geothermal Field

Lovelock, Cope and Baltasar

Table 1:

WATER AND STEAM CHEMISTRY OF TONGONAN WELLS UNDER THROTTLED CONDITIONS (HIGH WELLHEAD PRESSURE)
FOR SELECTED SAMPLING DATES DURING MEDIUM TERM DISCHARGE TEST

| WELL NO | DATE dd mm yy | WHP MPa | H° J/g | TMF kg/s | pH | Cl mg/kg | Na in water | K at | Ca atmospheric | S | SO ₄ pressure | SiO ₂ | Cl ₁ AQ (1) | Cl/B MOLEC | TSiO ₂ (2) | CO ₂ (3) | H ₂ S (3) | CO ₂ H ₂ S |
|---------|---------------|---------|-----------|----------|-----|----------|-------------|------|----------------|-----|--------------------------|------------------|------------------------|------------|-----------------------|---------------------|----------------------|----------------------------------|
| 103* | 11 08 78 | 2.74 | 1120 | 31 | 6.9 | 13100 | 6813 | 1947 | 234 | - | - | 981 | 8500 | - | 278 | 56 | 1.2 | 45 |
| 105 | 01 02 79 | 3.26 | 1220 | 45 | 7.1 | 12400 | 6375 | 1707 | - | 229 | - | 914 | 8000 | 16.5 | 273 | 61 | 2.7 | 22 |
| 108 | 12 06 80 | 3.25 | 1460 | 12 | 7.1 | 12090 | 6460 | 1680 | 231 | 267 | - | 972 | 8100 | 13.8 | 278 | 407 | 7.9 | 51 |
| 2R2 | 11 09 79 | 2.69 | 1650 | 25 | 7.2 | 12860 | 6948 | 1681 | 216 | 247 | 39 | 1096 | 7600 | 15.9 | 289 | 188 | 5.4 | 35 |
| 208A | 14 03 80 | 4.03 | 1490 | 10 | - | 12370 | 6601 | 1719 | 171 | 220 | 21 | 1082 | 7200 | 17.1 | 291 | 133 | 8.5 | 16 |
| 209A | 20 02 81 | 5.46 | 1550 | 9 | 6.2 | 13800 | 7120 | 2040 | 177 | 249 | - | 1191 | 8400 | 16.9 | 302 | 36 | 3.9 | 9 |
| 213 | 17 09 79 | 4.17 | 1310 | 45 | 6.8 | 13650 | 7147 | 1892 | 230 | 265 | 39 | 1135 | 8200 | 15.7 | 292 | 63 | 3.5 | 18 |
| 214 | 21 07 79 | 3.38 | 1310 | 38 | 7.4 | 11620 | 6246 | 1580 | 228 | 211 | 29 | 917 | 7900 | 16.8 | 273 | 122 | 2.0 | 62 |
| 303 | 15 07 80 | 1.82 | 1170 | 88 | 7.1 | 10600 | 5505 | 1410 | 273 | 187 | 31 | 859 | 7000 | 17.3 | 268 | 31 | 1.3 | 24 |
| 401 | 09 03 82 | 3.46 | 1700 | 49 | 6.6 | 16320 | 8888 | 2420 | 302 | 358 | 32 | 1021 | 10300 | 13.9 | 279 | 188 | 11.5 | 16 |
| 402* | 22 12 80 | 0.43 | 1620 | 6 | 4.0 | 6500 | 4041 | 617 | 58 | 152 | 2270 | 522 | - | 13.0 | 227 | 3856 | 73.0 | 53 |
| 403 | 08 01 81 | 2.51 | 1280 | 9 | 7.4 | 10800 | 5829 | 1393 | 183 | 207 | 22 | 975 | 6900 | 15.9 | 279 | 310 | 4.5 | 69 |
| 404 | 09 07 81 | 2.89 | 1430 | 39 | - | 12700 | 6802 | 1575 | 300 | 298 | - | 756 | 8200 | 13.0 | 258 | 437 | 4.8 | 91 |
| 405 | 17 12 80 | 2.49 | 1150 | 25 | 7.3 | 11880 | 6294 | 1462 | 275 | 252 | 33 | 797 | 8000 | 14.4 | 262 | 166 | 1.8 | 93 |
| 406 | 09 07 79 | 4.30 | 1600 | 23 | 6.9 | 14340 | 7245 | 1940 | 293 | 306 | 29 | 1229 | 8500 | 14.3 | 299 | 169 | 6.8 | 25 |
| 407 | 05 08 78 | 3.34 | 1510 | 25 | 6.8 | 16720 | 8539 | 2434 | 248 | 382 | 40 | 1169 | 9800 | 13.3 | 300 | 430 | 6.6 | 65 |
| 410 | 22 09 81 | 5.54 | 1740 | 20 | 6.6 | 17330 | 9130 | 2380 | 237 | 401 | 34 | 1350 | 9500 | 13.2 | ~320 | 281 | 11.6 | 24 |
| 501 | 16 01 79 | 3.2 | 1210 | 47 | 7.5 | 9066 | 4920 | 1249 | 133 | - | 17 | 915 | 6000 | - | 271 | 71 | 2.6 | 27 |
| 502* | 14 08 81 | 2.49 | 1040 | 11 | 7.9 | 4785 | 2594 | 674 | 61 | 69 | 36 | 885 | 3700 | 21.1 | 270 | 107 | 3.8 | 28 |
| 503 | 01 04 80 | 3.17 | 1250 | 19 | 7.4 | 11370 | 6079 | 1480 | 231 | 208 | 24 | 844 | 7300 | 16.7 | 266 | 93 | 1.9 | 50 |
| 504 | 10 09 80 | 1.90 | 1150 | 14 | 7.1 | 8116 | 4260 | 962 | 196 | 141 | 52 | 653 | 5800 | 17.6 | 246 | 47 | 2.1 | 23 |
| 5R7* | 12 02 82 | 0.38 | 1110 | 19 | 7.7 | 4740 | 2635 | 454 | 130 | 60 | 78 | 635 | 3200 | 24.1 | 244 | 48 | 1.7 | 29 |
| 508D* | 17 02 82 | 2.84 | 1030 | 19 | 7.4 | 9561 | 4578 | 1251 | 197 | 167 | 30 | 785 | 6700 | 17.5 | 261 | 57 | 1.1 | 52 |
| 509 | 21 12 81 | 1.68 | 1090 | 33 | 7.7 | 7345 | 3759 | 867 | 171 | 112 | 34 | 679 | 5000 | 20.0 | 249 | 73 | 1.2 | 60 |
| 5100 | 24 02 82 | 2.64 | 1120 | 12 | 7.6 | 7773 | 3798 | 972 | 161 | 127 | 31 | 812 | 5500 | 18.7 | 263 | 72 | 1.5 | 48 |
| 5110 | 10 04 82 | 3.72 | 1230 | 11 | 7.1 | 9673 | 4911 | 1314 | 176 | 167 | 21 | 939 | 6200 | 17.7 | 275 | 487 | 7.7 | 63 |
| MG1 | 09 05 81 | 3.07 | 1210 | 12 | 8.1 | 4065 | 2345 | 381 | 21 | 59 | 37 | 851 | 2700 | 21.0 | 267 | 218 | 1.6 | 134 |
| MG2D* | 09 12 81 | 2.59 | 982 | 38 | 8.1 | 4072 | 2311 | 439 | 28 | 60 | 45 | 791 | 2800 | 20.7 | 261 | 67 | 0.8 | 80 |
| TGE 4* | 01 11 75 | 0.65 | 815 | - | 8.2 | 3794 | 2194 | 239 | 86 | 35 | 90 | 378 | 3100 | 33.1 | 209 | ~17 | -0.2 | ~60 |
| HS#4 | 26 04 82 | - | (boiling) | - | - | 3868 | 2120 | 215 | 108 | 38 | 84 | 304 | - | 31.0 | 210 | - | - | - |
| HS#12 | 26 04 82 | - | (67°C) | - | 8.1 | 3602 | 1982 | 186 | 85 | 33 | 78 | 258 | - | 33.3 | 198 | - | - | - |

NOTES:

- WHP - WELLHEAD PRESSURE (1) Calculated Deep Aquifer Chloride Concentration (mg/kg); See Text.
H° - TOTAL DISCHARGE ENTHALPY (2) Silica Geothermometer Temperature (°C); From FOURNIER, R.O. and
TMF - TOTAL MASS FLOW ROWE, J.J. (1966). Corrected for excess discharge enthalpy.
HS - Bao Valley Hot Chloride Spring (3) Concentrations expressed as mmoles/100moles in total discharge.
* - Well not throttled during discharge test. ‡ Wells with anomalously low discharge enthalpies; see text.

homogeneity of the Bao Valley chloride springs and their close proximity suggests they have a common parent source.

Boron and arsenic in the well fluids are the toxic constituents of greatest concern and have made reinjection of separated wastewater a necessary part of the project development.

DOWNHOLE TEMPERATURE PROFILES

From an analysis of the temperature profiles measured in most wells in the Tongonan field isotherm contours have been developed for various depths of the field (Paete 1981). These contours have been found to resemble the surface contours of the diorite pluton as shown in Figure 1. However temperature inversions in some wells and particularly in the Malitbog sector have shown that the recorded temperatures do not directly correspond to inherent diorite rock temperatures but are more a result of reservoir fluid flowing from a higher temperature region.

GEOCHEMISTRY AND MEDIUM TERM DISCHARGE TESTS

Most of the geochemical data presented in this paper have been collected during a 6-12 week discharge of the wells, designed to give stable output and chemical data under both full bore discharge (FBD) and throttled conditions when the well head pressure (WHP) approaches the Maximum Discharge Pressure. After stable output is attained under FBD (low WHP) the well is throttled in one week

stages using successively smaller orifice plates. The chemical response to increasing WHP is found to be central to the understanding of well discharge processes.

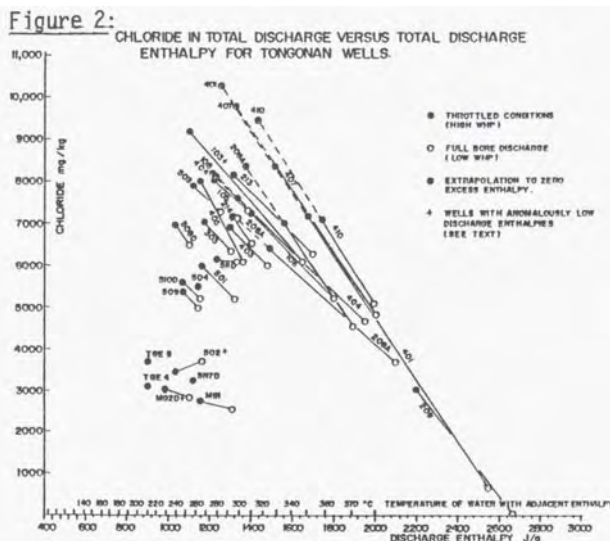
EXCESS ENTHALPY IN TONGONAN WELLS

Under FBD most wells in the Mahiao and Sambaloran sectors develop excess enthalpy, that is, total discharge enthalpies greater than the enthalpy of single-phase fluid at the reservoir temperature calculated from silica concentration. Many Tongonan wells have upper two-phase zones above the deep single-phase reservoir. Excess enthalpy is considered to be primarily derived from boiling in this upper zone, however in some of the hotter wells downhole surveys under flowing conditions have indicated two-phase conditions in the deep reservoir.

Tongonan wells exhibit highest discharge enthalpy under FBD and with throttling to higher WHP enthalpy usually declines to that of a single-phase fluid at the reservoir temperature. This is in contrast to some fields where vapor zones cause an increase in enthalpy at high WHP (Grant 1979). Some hotter Tongonan wells retain excess enthalpy even at high WHP while wells drilled in the cooler Malitbog reservoir generally do not develop significant excess enthalpies even under FBD.

ESTIMATING DEEP AQUIFER CHLORIDE CONCENTRATION

It is found for all high-enthalpy Tongonan wells that there is a consistent inverse linear



relationship between total discharge chloride concentration (Cl_{TD}) and total discharge enthalpy (H_{TD}). Figure 2 shows the linear regression lines of Cl_{TD} vs H_{TD} plots for all wells tested in Tongonan. This relationship can be explained in terms of dilution under FBD conditions by high enthalpy steam from a separate low-chloride two-phase feed, or alternatively by preferential entry of steam as boiling develops in the reservoir. In this paper the deep aquifer chloride concentration is obtained directly from the Cl_{TD} vs H_{TD} plot at the point where H_{TD} equals the reservoir temperature obtained from the silica geothermometer. This is the point where Cl_{TD} reaches a maximum and steam dilution is lowest. For the few wells that retain excess enthalpy even at higher WHP the Cl_{TD} vs H_{TD} plot is extrapolated to the single-phase condition. In Table 1 are presented the deep aquifer chloride concentrations calculated by the above method for all Tongonan wells that have undergone medium term discharge tests. The anomalously low measured enthalpies of some wells, based on silica and measured downhole temperatures, (Table 1), are considered unreliable and are not used in aquifer chloride calculations.

HYDROGEOCHEMICAL TRENDS ACROSS THE TONGONAN FIELD

In Figure 3 deep aquifer iso-chloride contours are mapped across the field. Aquifer chloride is found to be highest in the Upper Mahiao sector and declines in all directions away from this region. The most marked decline that has been observed is across the Malitbog sector, where on the southern periphery chloride approaches that of the TGE wells (approx. 3500 mg/kg).

Silica geothermometer temperatures (Fournier and Rowe 1966) are presented as contours in Figure 4. Well 410, the deepest well tested in the Upper Mahiao sector shows the highest silica temperature of 315-320°C. Temperatures are seen to decline to the west (Well 402), to the north (405), and to the east, (403) but remain high towards the Sambaloran sector. A more rapid decline is observed due south towards Well 504 (245°C) but temperatures remain considerably higher in the eastern Malitbog, (Well 511D, 276°C).

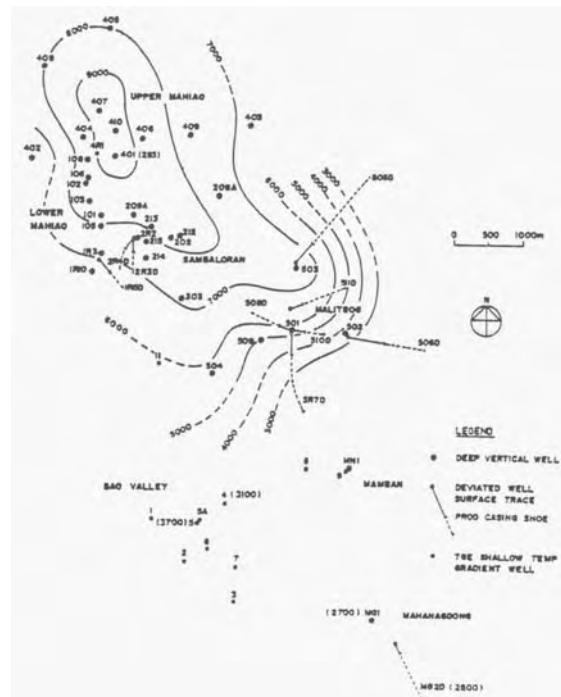


Figure 3: Deep Aquifer Chloride Concentration (mg/kg) Contours

An aquifer chloride versus silica temperature plot used for inferring dilution and heat transfer processes in the reservoir is shown in Figure 5. Although this graphical method has the limitation that a combination of processes can be used to invoke hydrological connections between most wells it is nevertheless useful for indicating more probable hydrological connections.

Three cooling-by-dilution trend lines are shown in Figure 5. Line A suggests a possible dilution pathway from Well 410 through 208A to 502. Line B suggests the shallow TGE wells 4 and 5 lie at the end of a dilution pathway through Wells 213, 508D and the southwestern Malitbog Wells 504 and 509. It appears less likely that the Bao Valley fluid is derived from eastern Malitbog Wells 502 and 5R7D since considerable conductive cooling or loss of steam would be required. Moreover Figures 3 and 4 suggest there are two separate outflows from the Sambaloran sector; to the east (Well 503) and to the south (Well 504).

It also appears that fluid does not flow directly from Well 501 to nearby 502 since the latter is considerably more dilute but at a similar temperature. Chloride and temperature trends support the view that the Malitbog sector is a major outflow zone and suggest further that it is comprised of several separated preferential fluid pathways.

Well 401 appears to have undergone considerable adiabatic boiling and cooling while wells along Line C in Figure 5 have probably been affected by a combination of boiling and dilution, (dilution alone, by 30°C water is not possible at reservoir depths).

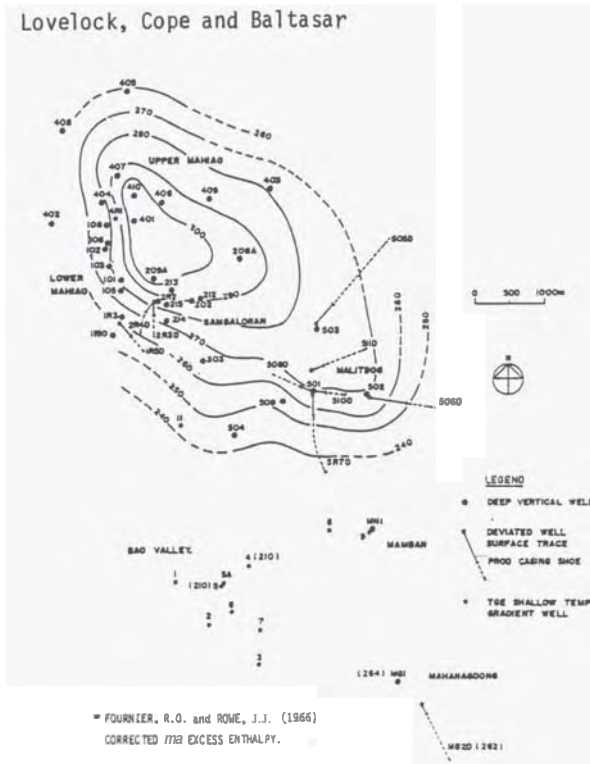


Figure 4: Silica geothermometer temperature (°C) Contours

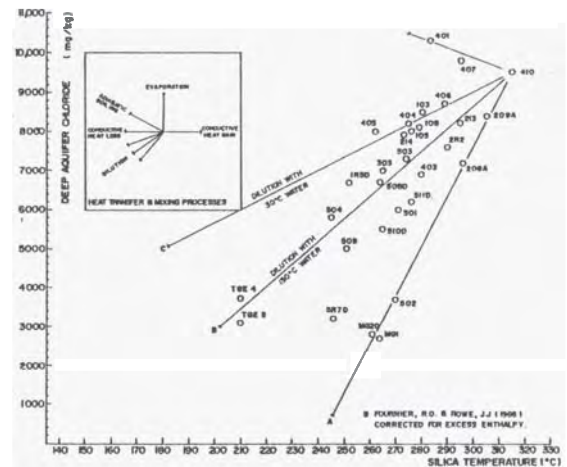
The reservoir fluid of the two Mahanagdong wells, MG1 and MG2D is considerably more dilute than that of other deep Tongonan wells and has a temperature of about 260°C, based on silica concentration. Based on this information and in view of the distance from the rest of the field, the Mahanagdong reservoir is considered to have a separate heat source from that of the northern sectors.

It can be seen from Figure 5 that for reservoir fluid to flow from the Mahanagdong sector to the Bao Valley considerable conductive cooling or loss of steam must occur. The overall geochemical trends suggest the Bao Valley chloride fluid is more likely derived from the north and the chemical homogeneity of the Bao Valley chloride springs makes the possibility of two widely-separated parent sources less likely.

GAS CHEMISTRY

Gas concentrations are known to be highest in the upper two-phase zones of Mahiao and Sambaloran wells, based on the following observations: Under discharging conditions and in the vicinity of the well bore these upper zones become underpressured with respect to the hot hydrostatic pressure curve and relative to the deep zone. At higher discharge pressures the relative contribution from the upper zone is less and for many wells (e.g.: 406, 403, 213, 208A) a marked decline in gas concentration is observed (together with a corresponding increase in total discharge chloride). Shallow wells 208, 209 had considerably higher gas levels before they were redrilled and deepened (208A, 209A). Well 403 showed highest gas levels early in the discharge when production was predominantly from shallower levels.

Figure 5: CALCULATED DEEP AQUIFER CHLORIDE VERSUS SILICA¹⁸ TEMPERATURE



Gas data are included in Table 1 for high well-head pressure conditions (when gas is generally lowest). Highest gas concentrations are found in Upper Mahiao wells and peripheral wells to the west (402, 408) and east (403, 208A). The very high-gas acid-sulphate fluid discharged by Well 402 appears related to the acid-sulphate Kapakuhan springs nearby, and suggests the incursion of acid-sulphate fluid derived from the near-surface oxidation of an H₂S-rich steam flow. Gas concentrations decline to the south across the Sambaloran and Malitbog sectors. Well 511D has anomalously high gas concentrations that may be transmitted upwards through narrow channels in the East Philippine Fault which is considered relatively permeable in this region. Well 511D is the first tested well to have intersected this fault.

The overall gas trends when considered in conjunction with the chloride and silica geothermometer data suggest that gas has migrated with separated steam away from the upwelling zone along preferential routes particularly to the west, Well 402, 408, and also to the east, Wells 208A, 403.

THE CHLORIDE/BORON TREND IN TONGONAN

The molecular Cl/B ratio increases consistently from a value of 14 in the Upper Mahiao to 25 in the southern Malitbog. Cl/B contours are shown in Figure 6. The south-trending increase has been recognised previously (Barnett 1979), and was interpreted to be due to depletion of boron in the deep aquifer through steam separation since boric acid is slightly volatile. This explanation may be valid in the Upper Mahiao and Sambaloran sectors and is in line with the south-trending decline in gas described above. Subsequently, it has been found that the greatest Cl/B increase is across the Malitbog sector where Cl/B tends to approach the Bao Spring Cl/B ratio of 32. Since the Malitbog reservoir is considered largely single-phase some other explanation must be invoked. The Cl/B ratio is often used to indicate the homogeneity of a geothermal reservoir since it is often similar to the ratio in the major aquifer rock type (Truesdell 1975) and should change little with subsequent



Figure 6: Molecular Chloride/Boron Ratio Contours

dilution. The upper Mahiao Cl/B ratio is consistent with a diorite rock type but the range of rock type observed across the field is not sufficiently marked to explain the Cl/B trend. The increase across the Malitbog and Sambaloran sectors and towards the Bao Springs may be due to loss of boron into the lattice structures of alteration clay minerals (Harder 1961).

STABLE ISOTOPE DATA

Water samples from rivers, hot springs and wells have been collected over the last 5 years for oxygen-18 and deuterium stable isotope analysis and this is more fully reported elsewhere. (Cope, et al., 1982). Tongonan geothermal fluids exhibit a maximum positive $\delta^{18}\text{O}$ shift of approximately 5.0‰ relative to the local meteoric water line; (the wells showing the greatest shift are 410, 407, 401, 209A) however the fluids have a higher deuterium content than local meteoric waters, (approx. 10‰ enrichment).

The deuterium enrichment of geothermal fluids with respect to local meteoric waters has been observed in other geothermal systems (Truesdell, et al., 1977) and has been explained in terms of subsurface boiling and dilution processes. For the purposes of this paper, a plot of aquifer chloride vs $\delta^{18}\text{O}$ has been constructed; Figure 7. Calculated aquifer chloride values have been used instead of total discharge chloride for wells with excess enthalpy except for well water samples that have obviously suffered dilution by either meteoric water (e.g. 402 (1), 202 (1) or drilling fluids (501). This is considered to be justified on the basis that steam separation which has occurred in the reservoir at temperatures above approximately

Figure 7:
DEEP AQUIFER CHLORIDE CONCENTRATION (mg/kg)
VERSUS $\delta^{18}\text{O}$ (‰)



280% (subsequently diluting the well discharge) will result in very little isotopic fractionation.

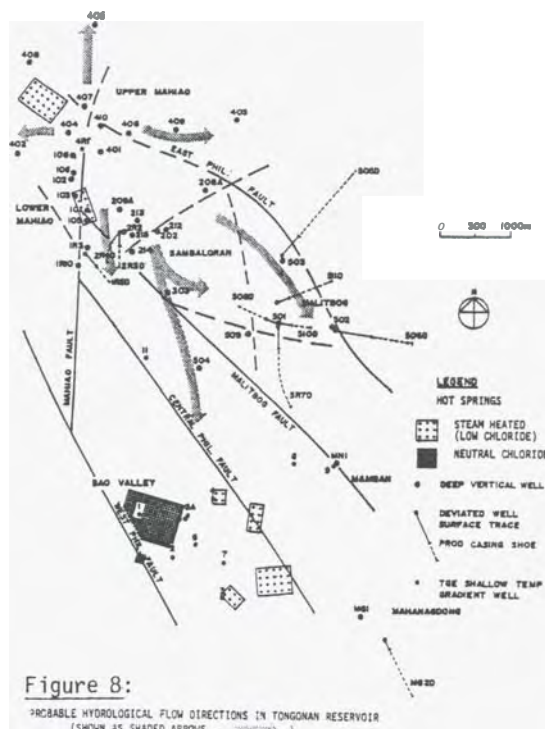
From Figure 7 two dilution lines have been drawn. The dilution line which passes through the Bao Valley thermal fluid composition supports the view of fluid flow to the Bao Valley via the western Malitbog, but the dilution water probably does not have the $\delta^{18}\text{O}$ composition (-3‰) implied by the line. The relative enrichment of Bao Valley thermal fluids in ^{18}O with respect to local meteoric water may be influenced by steam separation at temperatures of around 220°C and the diluting water is probably local meteoric water of composition; $\delta^{18}\text{O}$ -5.5‰ and δD ; -6.5‰.

Some wells appear to be diluted by a water of isotopic composition; $\delta^{18}\text{O}$; -5.5‰. and δD ; -30‰. which corresponds to the parent water for the geothermal fluids (in which no deuterium shift has occurred) but no local meteoric water with this isotopic composition has yet been found.

STRUCTURAL PERMEABILITY

Figure 8 shows the Tongonan field with thermal springs (Baltasar, et al., 1982), fault traces and probable reservoir flow paths. The geochemical, field trends have delineated the proposed major flow paths and some structural control of hydrology is evident. A consideration of the permeability of these structures is necessary in assessing their hydrological importance.

Primary permeability in the andesite and diorite rocks is low and fluid is believed to move through fractures formed by tectonic movement along faults and through intrusion of the diorite pluton. The present permeability of faults appears to vary greatly throughout the Tongonan field. Along the Central Philippine Fault low permeability due to



Although the Upper Mahiao region is considered to be the major upwelling zone in the Tongonan field, permeability (measured by injectivity during well completion tests) is not as high as that seen in the Sambaloran sector. However lower reservoir permeability is partly compensated for by the higher fluid temperatures which give water a lower viscosity. The Upper Mahiao wells also have lower total mass flows compared with Sambaloran wells. This is partly due to the higher enthalpies in the Upper Mahiao sector and the associated lower effective mass flux of steam through the reservoir compared to water.

Consistent field trends in geochemical parameters have been used to delineate probable hydro-

Dilution and cooling trends have been recognized in all directions away from the Upper Mahiao sector and for the deep wells so far drilled mineralization is lowest on the southern periphery of the Malitbog sector. The major outflow to the Bao Valley springs is considered to be derived largely from the western Malitbog area and possibly the Lower Mahiao.

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